

RNA Polymerase: Potent Linear Competitive Inhibition by  
D Arabinose-5-Triphosphate Compared to Non-inhibition  
by 5' Ara-ATP

D. Dennis, S. Jurgensen, J. Sylvester

Department of Chemistry, University of Delaware, Newark, Delaware 19711

Received March 19, 1980

ABSTRACT

D-arabinose-5-triphosphate is a potent linear competitive inhibitor ( $K_i = 60 \mu\text{M}$ ) for substrates of *E. coli* RNA polymerase [E.C.2.7.7.6] in an in vitro transcription system employing  $\Delta\text{D111}$  DNA as template. In contrast the corresponding nucleotide triphosphate (5' Ara-ATP) is without effect on the system. These facts coupled with other kinetic studies support the suggestion that the pentose triphosphate moiety of the normal substrates are bound at the catalytic site of the polymerase in the 3' endo ribose conformation and require the 2' hydroxyl in the axial position.

INTRODUCTION

Kinetic studies concerning the specificity of substrate and inhibitor binding interactions at the catalytic surface of the *E. coli* RNA polymerase have suggested that the conformation of the pentose moiety is critical. Tight binding of substrates or competitive inhibitors requires either a 3 hydroxyl of the 2 endo conformation or a 2 hydroxyl of the 3 endo conformation where the configuration of the interactive hydroxyl is the same configuration as ribose. Productive catalysis of the bound pentose moiety is suggested to be constrained to the 3 endo conformation where the 2 hydroxyl has the configuration of ribose. These initial studies (1,2) were reported for 5-ribose and 2 deoxy ribose polyphosphate analogues. This report is concerned with kinetic studies of the arabinose analogues,  $\beta$ ,D-arabinofuranosyl-5-triphosphate (Ara-5- $\text{P}_3$ ) and 9- $\beta$ ,D arabinofuranosyl 5' triphosphate (Ara -ATP).

Ara-ATP and other analogues (Ara-CTP and Ara-GTP) have been shown to be potent inhibitors of DNA polymerase from various sources (3-7). Ara-CTP and Ara-ATP have been reported to inhibit *E. coli* RNA polymerase (8,9) by acting as premature chain terminators. The presence of degradation products such as Ara-5- $\text{P}_3$  and Ara-5- $\text{P}_2$  are common in many commercial preparations and are difficult to remove except by HPLC purification. The presence of these degradation products and their effect on the kinetics of various enzymes may be a complication in the comparison of results. A possible mode of

therapeutic action in the case of Ara-NTP's may result from their degradation in vivo to Ara-5-P<sub>3</sub> or Ara-5-P<sub>2</sub> in which case they may function as potent inhibitors of RNA polymerase.

#### MATERIALS AND SYNTHESIS

Nucleoside triphosphates (HPLC purified) were obtained from I.C.N.; D arabinose-5-phosphate from Sigma Chemical Co. and [5,6 <sup>3</sup>H]UTP from New England Nuclear Corporation. The  $\Delta$ D111 (a T<sub>7</sub> mutant containing the A<sub>1</sub> promoter) DNA was extracted from  $\Delta$ D111 (obtained from J. Dunn, Brookhaven National Lab.) and purified through a cesium chloride gradient centrifugation. The RNA polymerase was isolated from *E. coli* K<sub>12</sub> and purified by the method of Zillig (10). The enzyme had a specific activity of 300 mU/mg and had an  $\sigma/E$  mole ratio of 0.6.

D arabinofuranosyl-5-triphosphate was synthesized from D arabinofuranosyl-5-monophosphate by the water soluble carbodiimide condensation employing H<sub>3</sub>PO<sub>4</sub> according to the method of Horecker, *et al.*, for the phosphorylation of D-ribose-5'-triphosphate from D-ribose-5-monophosphate (11). The product peak isolated from the Dowex-1-Cl chromatography gave a single spot on TLC (acetate backed cellulose sheets, solvent N-propanol, NH<sub>4</sub>OH, H<sub>2</sub>O; 50:20:25). The R<sub>f</sub> values were: Ara-5-P (0.84) and Ara-5-P<sub>3</sub> (0.41) when visualized with an anisidine/phthalic acid spray reagent. A sample of Ara-5-P<sub>3</sub> was subjected to mild acid hydrolysis and analyzed for labile phosphate (12) and orcinol positive pentose (13) with the following results: mole ratio of labile phosphate/orcinol pentose = 2.03. A sample of the mild acid hydrolysis was chromatographed (as above) and showed a single spot corresponding to Ara-5-P.

#### KINETIC ASSAYS

An assay reaction mixture was prepared so that when 80  $\lambda$  of the mixture was diluted to a final volume of 100  $\lambda$ , the reaction was: 40 mM tris-HCl (pH 7.9), 0.1 mM EDTA, 7.5 mM  $\beta$  mercapto ethanol, 50 mM KCl, 10 mM MgCl<sub>2</sub>, 0.4 mM in ATP, GTP, CTP and contained 1.2  $\mu$ g  $\Delta$ D111 DNA, and 0.56  $\mu$ g of *E. coli* holoenzyme ( $\sigma/E$  = 0.6). The final reaction mixture also contained [5,6 <sup>3</sup>H]UTP (12-100  $\mu$ M final concentration having 200 cpm/p mole) and either Ara-5'-ATP (at 0 or 50 or 500  $\mu$ M final concentration) or Ara-5-P<sub>3</sub> (12-100  $\mu$ M final concentration) as appropriate for a particular kinetic experiment. The reaction was initiated by combining the 80  $\lambda$  reaction mixture (preincubated at 37° for 5 minutes) with the 20  $\lambda$  increment containing UTP and the Ara derivative. The reaction was incubated for 10 minutes at 30°C and terminated by the addition of 1.0 ml of ice cold 10% trichloroacetic acid (TCA).

The samples were stored in ice until filtered through glass filter discs prewashed with cold 10% TCA. The discs containing the precipitate were washed three times each with 3.0 ml volumes of cold 10% TCA and finally with 2.0 ml of 95% ethanol, dried for 15 minutes at 100°C and then counted in a toluene based cocktail in a scintillation counter.

#### RESULTS AND DISCUSSION

The incorporation of [<sup>3</sup>H]UMP into acid insoluble polymer during a 10 minute incubation was measured as a function of UTP concentration (over a concentration range 12-100  $\mu$ M) with or without a constant concentration of Ara-5'-ATP (50 or 500  $\mu$ M) and analyzed. No inhibition was observed over this range of Ara-5'-ATP concentration.

The inhibition of the incorporation of [<sup>3</sup>H]UMP into acid insoluble polymer was also measured as a function of Ara-5-P<sub>3</sub> (over a concentration range 12-100  $\mu$ M) at different fixed concentrations of [<sup>3</sup>H]UTP. The data was analyzed

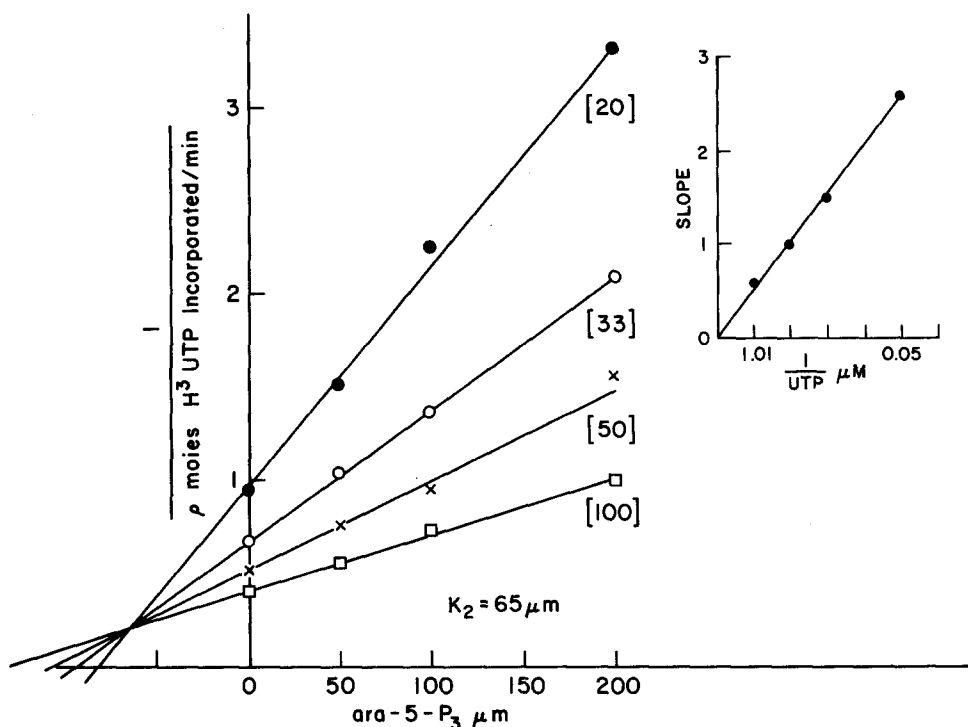
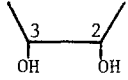
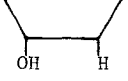
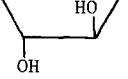
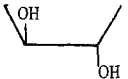
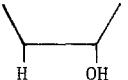
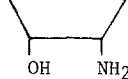
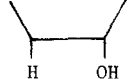


Figure 1. Inhibition by Ara-P<sub>3</sub>. a) Dixon plot where [ ] is the fixed  $\mu M$  concentration of UTP. b) Insert for slope (a) vs.  $1/[UTP]$ .

by plotting  $1/v$  vs.  $[Ara-5-P_3]$  at each concentration of  $[^3H]UTP$  as shown in Figure 1. A secondary plot of slope vs.  $[UTP]^{-1}$  was linear (14) therefore Ara-5-P<sub>3</sub> is a linear competitive inhibitor having a  $K_i = 60 \mu M$ .

The data presented in Table 1 support the hypothesis of Sylvester and Dennis (1) that the binding of the pentose polyphosphate moiety of nucleotide triphosphates (substrates as well as inhibitors) accounts for the major part of the interaction between these compounds and the *E. coli* polymerase. In detail the specificity of binding of the pentose polyphosphate moiety requires that either the 3 hydroxyl of the 2-endo conformation (see Figure 2), or the 2 hydroxyl of the 3-endo conformation, have the same configuration as ribose. Productive catalysis (formation of a phosphodiester bond between the bound 5 triphosphate and the pendant 3 hydroxyl of the growing RNA polymer) is suggested to be constrained to the 3-endo conformation where the 2 hydroxyl having the configuration of ribose is the interactive hydroxyl for binding. (For instance,  $[1-^{14}C]$  D ribose 5 triphosphate (2-endo) is not a substrate for the enzyme but is a potent competitive inhibitor.) The model assumes that the absence of the bulky base in position 1 of the pentose restricts the binding of the pentose-5-triphosphate series to the 2-endo conformation whereas the

Table 1. Kinetic Inhibition Studies Summary for Pentose-5-Triphosphates and/or Nucleopentose-5'-Triphosphates

D Pentose	2 and 3 hydroxyl conformation	$K_i$ $\mu$ M	
		Pentose- $P_3$	Nucleopentose- $P_3$
Ribose (1)		21	13 ( $K_m$ )
2d Ribose (1)		26	2800
Arabinose		60	>5000
Xylose (15)		--	37
3d Ribose (16)		--	18
2d 2 $NH_2$ Ribose (17)		--	>2000
3d 3 $NH_2$ Ribose (17)		--	2

presence of the bulky base in the nucleopentose-5-triphosphate series restricts the binding to the 3' endo conformation. Conformational restraints on binding are not to be confused with solution equilibria for these pentose conformations as a function of the position 1 substituents.

The D ribose-5-triphosphate is a potent competitive inhibitor as is 2d ribose-5-triphosphate since both possess the interactive 3 hydroxyl of the 2 endo conformation in the ribose configuration (axial). In contrast, 2'd

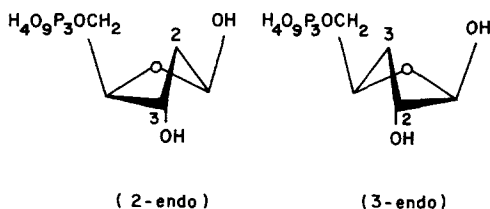


Figure 2. Two conformations of pentose moiety considered for binding at catalytic site.

nucleopentose triphosphate is neither a substrate nor effective inhibitor since the 3' endo conformation requires an axial 2' hydroxyl, and it is absent. Arabinose-5-triphosphate is an excellent competitive inhibitor having an axial 3' hydroxyl of the 2' endo conformation. In contrast, the Ara-ATP is not an effective inhibitor since the 3'-endo conformation does not possess an axial 2' hydroxyl. Xylo-ATP is again a potent competitive inhibitor since it possesses an axial 2' hydroxyl of the 3'-endo conformation. The 3'd ATP (cordicepin triphosphate) is not only bound tightly, but is also a substrate (functions as a chain terminator) since it is bound in the 3' endo conformation using its axial 2' hydroxyl. Consistent with the observation that ribose-5-diphosphate is a potent competitive inhibitor ( $K_i = 26 \mu\text{M}$ ), the diphosphate analogue of 3'd ribose (3d ADP) is not a substrate but is a good competitive inhibitor of the reaction ( $K_i = 37 \mu\text{M}$  in our studies,  $K_i = 106 \mu\text{M}$  in Bruzel (16)). 3d ADP is not a chain terminator as is the triphosphate. The 3' amino ATP is a good inhibitor of the enzyme whereas the 2' amino ATP is an ineffective inhibitor for the same reason as stated above.

The therapeutic effectiveness of Ara 5-NTP's has been suggested to be the inhibition of DNA polymerase at various stages and perhaps the inhibition of RNA polymerase. The presence of contaminants in non-HPLC purified commercial preparations of Ara-ATP and Ara-CTP (namely Ara-5-P<sub>3</sub> ~ 5-9%) indicate that caution be exercised in interpretation of the kinetic inhibition data. Our results with Ara-5-P<sub>3</sub> suggest that this agent might also be an effective therapeutic agent for neoplasias and viral infection in that transcription is effectively inhibited. Ara-5-P<sub>3</sub> may also prove to be an effective inhibitor of DNA polymerases.

#### ACKNOWLEDGEMENTS

This work was supported by a grant from the NSF, #PCM 7710226.

#### REFERENCES

1. Sylvester, S. and Dennis, D. (1977) BBRC 75, 667.
2. Dennis, D. and Sylvester, S. (1977) Fed. Proc. 36, 3221.
3. Cohen, S.S. (1966) Prog. Nucleic Acid Res. 5, 1.
4. Ono, K., Ohashi, A., Yamamoto, A. and Matsukage, A. (1979) Cancer Res. 39, 4673.
5. Brox, L., Mowles, D. and Belch, A. (1979) Eur. J. Cancer 15, 1263.
6. Fridland, A. (1977) Biochem. 16, 5308-5312.
7. Matsukage, A., Takahashi, T. and Nakayama, C. (1978) J. Biochem. 83, 1511.
8. Chuang, R. and Chuang, L. F. (1976) Fed. Proc. 35, 1109.
9. Chuang, R. and Chuang, L. F. (1975) Cancer Res. 35, 687.
10. Zillig, W., Zechel, K. and Halbwachs, H-J. (1970) Hoppe-Seyler Z. Physiol. Chem. 351, 221.
11. Horecker, B. L., Hurwitz, J. and Heppel, L. (1957) J.A.C.S. 79, 701.

12. Leloir, L. F. and Cardini, C. E. (1957) in *Methods in Enzymology*, Colowick, S. and Kaplan, N. O., Eds., Vol. III, pp. 840-850 Academic Press, New York, New York.
13. Putman, E. W. (1957) in *Methods in Enzymology*, Colowick, S. and Kaplan, N. O., Eds., Vol. III, pp. 70-71 Academic Press, New York, New York.
14. Segal, I. H. (1975) *Enzyme Kinetics*, p. 110, Wiley Interscience, New York, New York.
15. Dennis, D. and Thomas, W., unpublished data.
16. Bruzel, A., Suhadolnik, R. J. and Wilson, R. G. (1978) *Fed. Proc.* 37, 2012.
17. Armstrong, R. and Eckstein, F. (1976) *Eur. J. Biochemistry* 70, 33-38.